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PHYSICAL VS CHEMICAL PROPERTIES QUIZ

| 1. | Differentiate between physical and chemical properties. | | | | |
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| 2. | Classify the following as physical or chemical property by putting a tick in the appropriate box | | | | |

| Property | Physical | Chemical |
|-----------------------------------|----------|----------|
| | Property | Property |
| Denser than water | | |
| Supports combustion | | |
| Tastes bitter | | |
| Reacts with metals | | |
| Doesn't dissolve in water | | |
| Can neutralize an acid | | |
| The substance smells | | |
| The luster is bright | | |
| Can be bent without breaking | | |
| Produces hydrogen gas when heated | | |
| A white substance | | |

| 3. | Which of the following is a chemical property? |
|------------|--|
| Α. | Color |
| В. | State of matter |
| C. | Boiling point |
| D. | Corrosiveness |
| E. | Hardness |
| 4. | Which of the following is a physical property? |
| Α. | Flammability |
| В. | Reactivity |
| C. | Size |
| D. | Decomposition |
| 5 . | Which of the following statements describes a chemical property? |
| | |
| Α. | Copper objects turn green when exposed to moist air for a prolonged |
| | period. |
| В. | Aluminum metal is a solid at room temperature. |
| C. | Potassium metal reacts vigorously with water to form potassium hydroxide |
| | solution and hydrogen gas. |
| D. | The density of Lithium metal is 0.534g/ml at 20°C. |
| 6. | Which of the following statements is true? |
| Α. | Being a liquid at room temperature is a chemical property of Mercury. |
| В. | Resisting change when exposed to air for a prolonged period is a |
| | chemical property of Gold. |
| C. | Noble gases reacting less with the environment is a physical property of |
| | noble gases. |
| D. | Having a reddish-brown color is a chemical property of Copper. |

7. Circle the choice that best completes the statement or answers the

question.

- a. (Physical or Chemical or both) properties describe matter.
- b. (Physical or Chemical or both) properties of a substance can be easily observed.
- c. (Physical or Chemical or both) properties usually describe how a substance reacts.
- d. (Physical or Chemical or both) properties are not as easy to observe.
- e. (Physical or Chemical or both) properties can be used to identify substances.

b) By putting a tick in the appropriate box, identify if the following are chemical or physical properties:

| Property Description | Physical | Chemical |
|--|----------|----------|
| | Property | Property |
| Oxygen is odorless and colorless | | |
| Copper turns green when exposed to the | | |
| environment | | |
| The piece of metal is magnetic | | |
| Diamonds are a very hard substance | | |
| The piece of metal is magnetic | | |
| The tree is 8 meters high | | |
| Sodium reacts very easily with other elements. | | |
| A student's shirt suddenly bursts into flames. | | |
| The color of the ball is red | | |
| Iron reacts with oxygen and forms rust | | |
| The boiling point of water is 100 degrees C | | |
| The silver spoons tarnished and turned dark | | |
| Argon is not very reactive | | |