1 1 Most Common SEPARATION TECHNIQUES

Separates magnetic materials (such as iron) from nonmagnetic materials.

Use of magnets

02. **Filtration**

Separates an insoluble solid from a liquid or solution by passing it through a filter medium

Separates a mixture by carefully pouring off the liquid layer while leaving the solid sediment or precipitate behind.

03. Decantation

Separates a mixture by heating it, causing one

component to change

directly from a solid to a gas, leaving behind the





04. **Sublimation**

Separates liquids with different densities by allowing them to settle into distinct layers and then draining them through a tap.

05. Use of a Separating funnel

Separates solute from a

other components.



solution by cooling it, 06. prompting the solute to crystallize for easy Crystallization separation from the liquid.

Separates components of a mixture based on differences in their boiling points by heating the mixture, then collecting the condensed vapors.

Simple Distillation





08. **Fractional Distillation**

Separates components of a mixture based on their affinity for a stationary phase and a mobile solvent.

Separates mixture components with closer boiling points more effectively than simple distillation by employing a fractionating column for enhanced separation efficiency.

09. Chromatography





10. Solvent Extraction

Separates components of a mixture based on differences in their density by spinning the mixture at high speeds.

Separates substances based on their solubility in two immiscible solvents, usually water and an organic solvent.

11. Centrifugation

